

THE OPEN UNIVERSITY OF SRI LANKA B.Sc. DEGREE PROGRAMME - 2019/2020 LEVEL 4 - CYU4300 INORGANIC CHEMISTRY ASSIGNMENT TEST II (NBT)



4.15 p.m. − 5.15 p.m. DATE: 3rd September 2019 Avogadro constant, L = $6.023 \times 10^{23} \text{ mol}^{-1}$ Velocity of light, c = $3 \times 10^8 \text{ m s}^{-1}$ $= 1.661 \times 10^{-27} \text{kg}$ 1 a.m.u. $= 1.6021 \times 10^{-13} J$ 1 MeV Answer all questions Mark a cross X over (English letter) that corresponds to the most suitable answer on the given answer sheet. Any answer with more than one X will not be counted. 1. Consider a part of the decay series given below. Which of the following statements are true? $^{238}_{92}U \xrightarrow{-\alpha} X \xrightarrow{-x} ^{234}_{91}Pa \xrightarrow{-\beta^{r}} Y \xrightarrow{206}_{82}Pb$ (i) It is (4n+2) decay series (ii) X is $^{234}_{99}Th$ (iii) x is β^- (iv) Y is $^{233}_{99}Th$ The answer is c) (iii) and (iv) only b) (ii) and (iii) only a) (i) and (ii) only e) (i), (ii) and (iii) only d) (i) and (iv) only 2. How does $^{218}_{84}Po$ decay to $^{218}_{85}At$? a) By positron emission b) By electron capture c) By electron emission d) By neutron emission e) By α-decay 3. At 12.00 noon, in a nuclear pharmacy, the activity of the radioactive indium-111 was found to be 10 mCi. What will be the activity of indium-111 in mCi at 13.30 hrs the same day? The half-life $(t_{1/2})$ of indium-111 is 2.83 days. c) 8.95 d) 9.85 b) 8.75 4. The activity of 1 μ g of pure indium-111 ($t_{1/2}$ = 2.83 day) in Becquerel (Bq) is b) 1.54×10^7 c) 1.54×10^4 d) 1.32×10^7 a) 1.54×10^{10} e) 1.32×10^4 5. Which of the following nuclides will be expected to be unstable and radioactive. (d) $^{20}_{10}Ne$ (e) $^{40}_{20}Ca$ $(c)^{-16}O$ (b) ${}_{6}^{11}C$ (a) ${}_{2}^{4}He$ 6. What are the modes of decay that ${}^{13}_{7}N$ may undergo? (ii) electron emission (i) y emission (iii) positron emission (iv) electron capture The answer is c) (iii) and (iv) only b) (ii) and (iii) only a) (i) and (ii) only

e) (i), (ii) and (iii) only

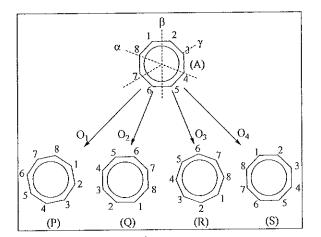
d) (i) and (iv) only

7. What	will be the p	roduct fo	ormed when 15	O undergo	es electron	capture?	ı
8	a) ${}_{9}^{15}F$	b) 16 F	c) $^{16}_{8}C$	d)) 14 N	e) ¹⁵ ₇ N	
(a) ele	ectron captur	e	that ${}^{22}_{9}F$ is lik (b) positron e (e) γ emission	mission	-	ectron em	nission
9. Identi	fy x in the nu	ıclear rea	action given by	the notati	on, ${}_{5}^{10}B(x, a)$	$(\alpha)_3^7 Li$:	
а) α	b) <i>n</i>	c) β ⁻	d)	β^+	e) γ	
$^{235}_{92}U$ (i) f	$+ \frac{1}{0}n \rightarrow \frac{146}{54}$ ission	$Xe + \frac{9}{38}$	true about the ${}^{4}Sr + 2{}^{1}_{0}n$ tron capture				iin reaction
The ans a) d)	wer is (i) and (ii) c	only	b) (ii) and (iii) e) (i), (ii) and	only) and (iv)	
(i) T a (ii) ((iii) ((iv) T	The method is tmospheric nearbon-14 un The materials The error involver is	based o itrogen v dergoes suitable olved in	for radiocarbothe determinat	on of <i>carbo</i> on dating a ion of age	on-14 by in re metals a of a specin	radiation nd glass nen is ±10	of 00 year
a) d)	(i) and (iv) o	niy only	b) (ii) and (iii) e) (i), (ii) and	only (iii) only	c) (111 _.) and (iv)	only
per m activ (t _{1/2} c	iinute (dpm). ity of 3.85 dp if <i>carbon-14</i>	If a pied m per gr is 5730 y	ng tissue has a be of charcoal; ram of carbon, /).	from a prel the age in	historic site years of th	is found e campsit	to show an
2.014 in the	1017, 4.002 nuclear fusi	6033 and on, ${}^{3}_{1}H + {}^{2}_{1}H$	of ${}_{1}^{3}H$, ${}_{1}^{2}H$, ${}_{1}^{3}H$, ${}_{1}^{3}H$, ${}_{1}^{3}H$, ${}_{1}^{4}H$, ${}_{1}^{4}H \rightarrow {}_{0}^{1}n + {}_{2}^{4}H$	spectively	, the energy	y released	l (MeV)
ŕ			c) 18.8 n in answerin	g question	·	5 6	e) 8.8

A configuration of the planar aromatic ion, $C_8H_8^-$, (on the paper) is shown in figure (A). Axis α passes through two opposite C nuclei. Axis β passes through two opposite C-C bonds bisecting them. Axis γ passes through the centre of the ion and is perpendicular to the plane of the ion. Figures (P), (Q), (R) and (S) show the resultant configurations of $C_8H_8^-$ when operations, O_1 , O_2 , O_3 and O_4 are performed on

configuration (A). The carbon nuclei labels are also shown. [All the C-C bond lengths and all the C-C-C bond angles in $C_8H_8^-$ are equal.]

- 14. Consider the following statements.
 - (i) Configurations (A) and (Q) are equivalent.
 - (ii) Configurations (P) and (R) are identical.
 - (iii) Configurations (Q) and (S) are equivalent.



The correct statements, out of (i), (ii) and (iii) above are,

- (a) Only (i) and (ii).
- (b) Only (i) and (iii).
- (c) Only (ii) and (iii).

- (d) All (i), (ii), and (iii).
- (e) None of the answers, (a), (b) (c) or (d) is correct.
- 15. Consider the following statements.
 - (i) Operation O_1 is a symmetry operation of $C_8H_8^-$.
 - (ii) Operation O₂ is a symmetry operation of C₈H₈⁻.
 - (iii) Operation O₄ is the identity operation of C₈H₈.

The correct statement/s is/are,

- (a) Only (i) and (ii).
- (b) Only (i) and (iii).
- (c) Only (ii) and (iii)

- (d) All (i), (ii), and (iii).
- (e) Only (iii).
- 16. Consider the following statements.
 - (i) Axis α is a symmetry element of $C_8H_8^-$.
 - (ii) The compound operation of O_2 followed by operation O_4 is a symmetry operation of $C_8H_8^-$.
 - (iii) Axis β is a symmetry element of $C_8H_8^-$.

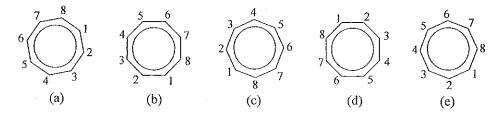
The correct statement/s is/are,

- (a) Only (i) and (ii).
- (b) Only (i) and (iii).
- (c) Only (ii) and (iii).

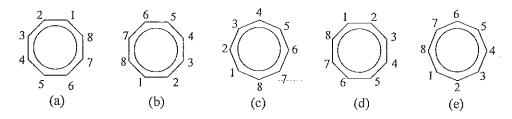
- (d) All (i), (ii), and (iii).
- (e) Only (ii).
- 17. Which of the following best represents the order of the rotational symmetry axis, γ , of $C_8H_8^-$?
 - (a) 4
- (b) 5
- (c) 6
- (d) 7

(e) 8

18. Which of the following best represents resultant configuration when operation O_4 is performed on configuration (R) of $C_8H_8^-$?

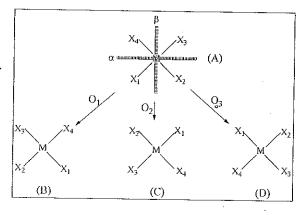


19. Which of the following best represents resultant configuration when configuration (A) is rotated by 180° about the axis β ?



Use the following information in answering questions 20 and 21.

Figure (A) shows a configuration of a square planar complex ion, MX_4^{2-} (on the paper). α and β are two mirror planes which are perpendicular to each other and perpendicular to the plane of the paper. Each mirror plane bisects two XMX angels as shown in the figure. Figures (B), (C), and (D) represent the resultant configurations after performing the operations, O_1 , O_2 and O_3 , respectively.



- 20. Consider the following statements.
 - (i) All three operations, $\rm O_1$, $\rm O_2$ and $\rm O_3$, are symmetry operations of $\rm MX_4^{2-}$.
 - (ii) Operation $\,O_1\,$ can be a reflection operation through $\,\beta\,.$
 - (iii) Performance of operation O₃ on (B) gives (C).

The correct statement/s is/are,

- (a) Only (i) and (ii).
- (b) Only (i) and (iii).
- (c) Only (ii) and (iii).

- (d) All (i), (ii), and (iii).
- (e) Only (i).
- 21. Consider the following statements.
 - (i) Planes, α and β are symmetry elements of $MX_4^{2-}.$
 - (ii) You <u>cannot</u> transform configuration (C) to configuration (D) through a symmetry operation of MX_4^{2-} .
 - (iii) The plane of the ion is a symmetry element of MX_4^{2-} .

The correct statement/s is/are,

- (a) Only (i) and (ii).
- (b) Only (i) and (iii).
- (c) Only (ii) and (iii).

- (d) All, (i), (ii), and (iii).
- (e) Only (ii).

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	and the second of a second of
22. Consider the following states molecule, NH ₃ .	tements about the symmetry elements of an ammonia
(i) Its principal axis is C ₃	
(ii) It has 3 vertical planes	
(iii) It has 3 dihedral plane	es of symmetry.
The correct statement/s is/are, (a) Only (i) and (ii).	(b) Only (i) and (iii). (c) Only (ii) and (iii).
(d) All (i), (ii), and (iii).	(b) Only (i) and (iii). (c) Only (ii) and (iii). (e) None of the <u>answers</u> , (a), (b) (c) or (d) is correct.
23. A student suggested the following reflection operations about CH ₄ .	llowing relationships (in standard notation) between any of the symmetry planes of a methane molecule,
	$\Xi \text{(iii) } \sigma = \sigma^3 \text{(iv) } \sigma^2 = \sigma^4$
The correct relationships are,	(b) Only (iii) and (iv) (a) Only (i) (ii) and (iii)
(a) Only (i) and (ii). (d) Only (ii), (iii) and (iv).	(b) Only (iii) and (iv).(c) Only (i), (ii) and (iii).(e) All (i), (ii), (iii) and (iv).
(i) Some molecules may h (ii) If there is a single nuccentre, if exists, is located	the inversion centre appear in pairs, which exchange
(a) Only (i) and (ii).	(b) Only (i) and (iii). (c) Only (ii) and (iii).
(d) All (i), (ii), and (iii).	 (b) Only (i) and (iii). (c) Only (ii) and (iii). (e) None of the <u>answers</u>, (a), (b) (c) or (d) is correct.
in its <i>staggered</i> configurate (i) It is a C_3 axis.	
(ii) Rotation of the moloperation of the moloperation (iii) It is an S ₃ axis.	lecule by an angle 480^{0} about it, is a symmetry lecule.
The correct statements are,	
(a) Only (i) and (ii). (d) All (i), (ii), and (iii).	(b) Only (i) and (iii).(c) Only (ii) and (iii).(e) None of the <u>answers</u>, (a), (b) (c) or (d) is correct.

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THE OPEN UNIVERSITY OF SRI LANKA

B. Sc. Degree Programme - Level 4 CAT-II - 2019/2020

CYU4300 - Inorganic Chemistry

MCQ Answer Sheet: Mark a cross (\times) over the English Letter that corresponds to the most suitable answer.

Reg. No.	

FOR EXAMINER'S USE ONLY				
Answers	No.	Marks		
Correct				
Wrong				
Total				

c a d c d 10 а b 11 c d 12 13 14 d 15 a c 16 17 d 18 19 b ď c 20 ď 21 d 22 a b c b q 23 c 24 d



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Answer Guide for CAT-II-2019/2020 CYU4300 – Inorganic Chemistry held on 03-09-2019

MCQ ANSWERS

1. e 2. c 3. d 4. a 5. b 6. c 7. e 8. c 9. b 10. d

11. a 12. c 13. d 14. b 15. c 16. d 17. e 18. e 19. a 20. d

21. b 22. a 23. d 24. c 25. a

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