The Open University of Sri Lanka Faculty of Natural Sciences Diploma in Food Science Programme



Department : Chemistry

Name of the Examination : Final Examination

Course Code and Title : CYD3612 Introduction to food science and

constituents of food

Academic Year : 2020/2021

Date : 18/09/2022

Time : 09.30a,m.-12.30p.m.

Duration : 2 hours

Index number

General Instructions

- 1. Read all instructions carefully before answering the questions.
- 2. This question paper consists of Four questions in six pages.
- 3. Answer All FOUR (04) questions. All questions carry equal marks.
- 4. Answer for each question should commence from a new page.
- 5. Draw fully labelled diagrams where necessary
- 6. Having any unauthorized documents/ mobile phones in your possession is a punishable offense
- 7. Use blue or black ink to answer the questions.
- 8. Circle the number of the questions you answered in the front cover of your answer script.
- 9. Clearly state your index number in your answer script.

1.	A) Food components can be classified into two groups, essential and non-essential
	depending on whether they could be biosynthesized in the body.

- i) Define the terms essential and non-essential food components.
- ii) Give two examples each for essential and non-essential food components.
- iii) Nutrients composition of food govern by several factors. Give three (03) factors of them.

(35 marks)

- B) Food Technology is the science that deals with all the technological and engineering aspects related to Food processing to marketing.
 - i) What is meant by food processing?
 - ii) Why is food preservation necessary for food technology?
 - iii) State three preservation techniques used in the early days.

(30 marks)

- C) Water is an essential component in any food item.
 - i) Draw the structure of water and give the H-O-H bond angle.
 - ii) Name the hybridization found in water molecule.
 - iii) name two major sources of water found in our body.

(35 marks)

- 2. A) Carbohydrates are essential components of food.
- i) Classify the following sugars as monosaccharides, disaccharides, and polysaccharides.
 - a) Glucose
 - b) Starch
 - c) Maltose
 - d) Sucrose
 - e) Lactose
- ii) Give three (03) uses of carbohydrates in our body.

(30 marks)

B) Maltose is a disaccharide of α -D-Glucose. The structure of α -D-Glucose is given below.

α -D-glucose

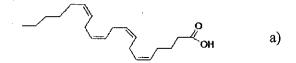
- i) Show the monomeric carbon of α -D-Glucose.
- ii) Explain why the above structure is named as α Glucose.
- iii) Draw the structure of maltose and indicate the glycosidic bond.

(20 marks)

- C) Carbohydrates undergo different chemical reactions.
 - i) Explain the classification of sugars as reducing and non-reducing, giving one (01) example for each.
 - ii) A laboratory technician in a Food laboratory receives an unknown sugar sample X, which is galactose or starch. He did the following tests, and the observations are given below. Identify the sugar sample with a suitable explanation for each sugar sample . X gave a violet colouration at the interface obtained by adding concentrated sulphuric acid down the sides of the tube to a solution heated with α -naphthol. X gives a silver mirror on heating with Tollen's reagent in a water bath. It did not provide blue colouration with iodine.
 - iii) Briefly explain the chemical reaction taking place during the Tollen's test. (50 marks)

3. (A)

i) Giving reasons, name the shorthand notation according to the ω_x nomenclature of the following fatty acids.



$$\mathsf{HO} \bigvee_{\mathsf{O}} \bigvee_{\mathsf$$

- ii) The smoke point of butter is 150°C, whereas the smoke point of coconut oil is 204
- °C. Which is the most suitable for deep frying of meat? Explain your answer.
- iii) Draw the structure of the following fatty acid molecules.
- a) Lauric acid (12:0)
- b) Palmitoleic acid 16:1 (n-7)

(44 marks)

- (B) i) Name three factors responsible for the solubility of fatty acids.
 - ii) Write down three (03) main sources of vitamin D.
 - iii) Give two (02) diseases that cause a deficiency in vitamin D.
 - iv) Name two (02) functions of vitamin A.

(36 marks)

- (C) The trans isomer of oleic acid (18:1 (n-9)) is called Elaidic acid. Elaidic acid is formed during the partial hydrogenation of vegetable oils.
 - i) Explain why oleic acid melts at 4 °C, but Elaidic acid melts at 45°C.
 - ii) Compare how the consumption of oleic acid and Elaidic acid affects the level of HDL and LDL cholesterol.

(20 marks)

- 4. A) i) What do you mean by dietary minerals?
- ii) Write down two (02) enhancing factors and inhibition factors of calcium absorption in the human body.
 - iii) What do you mean by goitrogenic food? Give two examples.
 - iii) Write down three (03) ways of fortification with minerals?

(31 marks)

- B) i) Briefly explain the reason for colour changes of fresh meat from bright red to pink colour during processing.
 - ii) Write down two (02) possible texture changes that occur when cooking plant-based food material.
 - iii) Name three compounds that impart a bitter taste in food.

(45 marks)

- (C) Briefly explain the reasons for the followings.
 - i) Consuming large amounts of raw eggs will cause biotin deficiency
 - ii) Vitamin C is an unstable vitamin that is readily lost during food processing and storage.
 - iii) Iron in plant food is not readily available, while iron in animal sources is readily available.

(24 marks)