

Reg. No.					

The Open University of Sri Lanka

B. Sc. Degree Programme - 2014/2015

**Faculty of Natural Sciences** 

**Department of Chemistry** 

**CMU3124** - Chemistry of Biomolecules

CONTINUOUS ASSESSMENT TEST I

Ques No.	Marks
1	
2	
3	
4	
5	
Total	

Date: Sunday, 12<sup>th</sup> July 2015

Time: 2.30 p.m. – 3.30 p. m.

1. (i) Draw the cyclic structures of D-glucose and D-fructose.

CHO 
$$CH_2OH$$
 $-OH$ 
 $-OH$ 

(10 Marks)

(ii) Draw the structure of sucrose, formed from glucose and fructose.

2. (i) Draw the structure of nucleotide formed from 2-deoxyribose, adenine and phosphate.

(10 Marks)

(ii) Draw the predominate form of glutamic acid at pH = 0, 3, 8, 11 given that pI = 3.22,  $pka_1 = 2.19$ ,  $pka_2 = 9.67$  and  $pka_3$  (side chain) = 4.2.

COOH
$$H_2N \longrightarrow C \longrightarrow H$$

$$(CH_2)_2COOH$$
glutamic acid

pH = 0

pH = 3



pH = 8

pH = 11

(20 Marks)

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3. (i) Write the structural formula for the tripeptide Val.Ala.Gly.

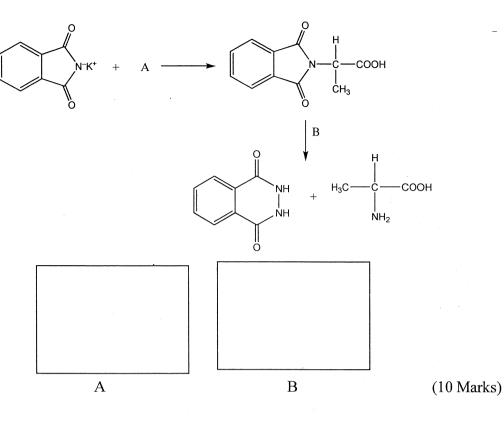
$$H_2N$$
— $CH$ — $COOH$   $H_2N$ — $CH$ — $COOH$   $H_2N$ — $CH_2$ — $COOH$   $H_2N$ — $CH_3$   $CH(CH_3)_2$  Alanine  $Valine$   $Glycine$ 

(10 Marks)

(ii) 2,4-Dinitroflurobenzene is used for N-terminal analysis of peptides. Write down the reactions involved when the tripeptide Val.Ala.Gly is treated with 2,4-dinitroflurobenzene and then hydrolysed with 6 N HCl.

4. Indicate the missing reagents and products in the following synthetic schemes.

(i)



5. What is the structure of the reaction product obtained when the tripeptide (E) is treated with phenylisothiocyanate followed by mild acid hydrolysis? Write down the reaction involved and explain the advantage of the reaction used to determine the N-terminal amino acid.

(20 Marks)



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## Continuous Assessment Test I - Answer Guide

1. (i)

(ii)

2. (i)

(ii)

$$\begin{array}{c|c} & & & & & & & & & \\ \hline & & & & & & & & \\ H_3N^+-C-H & & & & & & \\ (CH_2)_2COOH & & & & & & \\ pH=0 & & pH=3 & & & \\ \hline & & & & & & \\ PH=0 & & pH=3 & & & \\ \hline & & & & & & \\ COO^- & & & & & \\ H_3N^+-C-H & & & & \\ (CH_2)_2COO^- & & & & \\ (CH_2)_2COO^- & & & \\ pH=8 & & pH=11 & \\ \hline \end{array}$$

Since, the hydrolysis conditions are mild, only the terminal amino acid derivative is cleaved, and the rest of the peptide chain is intact. It is then recycled through further Edman degradation in an automated cycle that permits the successive cleavage and identification of a sequence of AA residues.